

Stereocontrolled Preparation of *cis*- and *trans*-2,6-Dialkylpiperidines *via*Diastereoselective Reaction of 1-Aza-4-oxabicyclo[4.3.0]nonane Derivatives with Grignard Reagents

Hadi Poerwono,[‡] Kimio Higashiyama,* Takayasu Yamauchi, Hajime Kubo, Shigeru Ohmiya, and Hiroshi Takahashi

Institute of Medicinal Chemistry, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142-8501, Japan

Received 15 July 1998; accepted 7 September 1998

Abstract: We report here the syntheses of cis- and trans-2,6-disubstituted piperidines using chiral 1-aza-4-oxabicyclo[4.3.0]nonane synthon 1, which shows high reactivity toward nucleophilic attack at its C-5 position. Bicyclic compounds resembling synthon 1 were transformed to cis- and trans-2,6-disubstituted piperidine derivatives via reactions with various Grignard reagents in a stereospecific manner. Using this methodology, (+)-solenopsin A (2b) and both enantiomers of isosolenopsin A (3a and 3b) were synthesized in an enantioselective manner from a single enantiomeric source. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: Bicyclic heterocyclic compounds; Grignard reactions/reagents; Piperidines; Solenopsin A.

1. INTRODUCTION

There has been significant interest in solenopsins (i.e., solenopsin A (2a), B and C along with the corresponding *cis* derivatives isosolenopsin A (3a), B and C), the major components of the non-proteinaceous venom of the fire ant *Solenopsis* (Myrmicinae), ever since their structures were determined by MacConnell and co-workers in 1970. These fire ant alkaloids exhibit cytotoxic, hemolytic, necrotic, antibacterial, insecticidal and antifungal activities. Their *cis-trans* isomerism, combined with their simple structures and bioactive potency, make them an ideal target for testing the stereoselectivity of numerous reactions. Although the absolute stereochemistry of natural solenopsins was not unambiguously resolved until Braekman and co-workers confirmed the *R*-configuration of the methyl moiety in 1994, there are currently 30 different synthetic routes for solenopsins, 10 of which exist in optically active forms. The methods used to prepare the parent 2,6-disubstituted piperidine ring structure include catalytic or chemical reduction of the corresponding pyridine derivatives, intramolecular cyclization of olefins or aminoketones, and cycloaddition. Beckmann rearrangement of oxime sulfonates, and cycloaddition.

^{*} Corresponding author; fax: 81-03-5498-5768; e-mail address: kimio@hoshi.ac.jp

[‡] On leave from: Airlangga University, Faculty of Pharmacy, Surabaya 60286, Indonesia

Chart 1

In a recent publication, ¹⁴ we described the syntheses of 2-substituted and 2,6-disubstituted piperidines using the chiral 1-aza-4-oxabicyclo[4.3.0]nonane synthon 1, which shows high reactivity toward nucleophilic attack at its C-5 position. Stereocontrol over both of the α positions of the piperidine derivatives depends on both the stereochemistry and the steric nature of the C-2 and C-9 substituents of 1. Although this procedure has been successfully applied to the syntheses of enantiopure 2-alkyl- and *cis*-2,6-dialkylpiperidine alkaloids, such as the α-pipecoline enantiomers, ¹⁴ (-)-sedamine, ¹⁵ and pinidine enantiomers, ¹⁶ the synthesis of the respective *trans* isomers gave poor results due to poor stereoselectivity. To investigate the role of *cis-trans* isomerism in the preparation of *cis*- and *trans*-2,6-dialkyl piperidines with a reasonable yield, we have extended our previous work to the reaction of bicyclic compounds resembling synthon 1 with various Grignard reagents in a stereospecific manner. Using this methodology, unnatural (+)-solenopsin A (2b) and both enantiomers of isosolenopsin A (natural 3a and unnatural 3b) were synthesized in an enantioselective manner.

2. RESULTS AND DISCUSSION

In a preliminary study, ¹⁴ both of the C-9 diastereomers of synthon 1 were prepared in three steps from the readily available, enantiomerically pure amino alcohol (*R*)-*N*-(2,4,6-trimethoxybenzyl)phenylglycinol (4), which acts as a chiral auxiliary as depicted in Scheme 1. As the C-9 substituents, we chose a methyl group to represent an aliphatic substituent and a phenyl group to represent an aromatic substituent. The stereochemistry at C-9 is established through the chiral 1,3-oxazolidine derivative intermediate, ¹⁷ which can provide the required stereoselectivity. By changing the functional groups of both the aldehydes (5 and 8) and Grignard reagents (6 and 9), the desired compounds 7a,b and 10a,b were obtained, with some of them as inseparable mixtures of C-5 epimers.

Scheme 1^s

As part of our initial studies, the asymmetric reactions of bicyclic compounds 7 and 10 were carried out by adding three equivalents of Grignard reagents to a THF solution of the bicyclic compounds. Compounds 7a and 10a gave the *cis* and *trans* isomers of piperidine derivatives, respectively, upon reaction with phenylmagnesium bromide. However, both 7b and 10b predominantly gave the *cis* isomer of piperidine derivatives in the reaction with methylmagnesium bromide. The preceding experiment and the current effort to achieve high stereoselectivity of the *trans* isomer, by reacting bicyclic 10b with various Grignard reagents, as well as the stereoelectronic mechanism are decribed below.

2.1. Asymmetric reaction of synthon 7a,7b with Grignard reagents

First, the asymmetric reaction of 7a with phenylmagnesium bromide was performed to predominantly give the *cis* isomer 11a, a piperidine derivative with a phenyl moiety at both of its α -carbons, in excellent yield, as depicted in Table 1. Similarly, the bicyclic synthon 7b with a methyl group at C-9 was treated with methylmagnesium bromide to completely give the cis- α , α '-dimethylpiperidine derivative (14) in moderate yield. We then reacted bicyclic compound 7a with methylmagnesium bromide to give almost exclusively the cis isomer 12a, and reacted 7b with phenylmagnesium bromide to give only the cis isomer 13, both in excellent yield.

This outcome reveals the mechanism involved in controlling the stereoselective attack of the methyl or phenyl function of the Grignard reagents toward the C-5 position of the starting bicyclic 7a or 7b. As shown in Table 1, the C-5 diasteromerically pure bicyclic 7a give a mixture of the *cis-trans* isomers of the piperidine derivatives 11a+11b (82:18) and 12a+12b (97:3). On the other hand, the C-5 diastereomeric mixture of 7b give only the *cis* isomer of the piperidine derivatives 13 and 14. The diastereoselectivity of this reaction can be

explained by assuming that the Grignard reagent approaches the oxygen atom of the 1,3-oxazolidine ring of the bicyclic compound to give a favorable intermediate iminium salt, as suggested in the reaction of 1,3-oxazolidine itself with a Grignard reagent. ^{16,17}

Table 1. Stereoselective Reactions of 7a/7b with Grignard Reagents^a

Substrate	Grignard reagent ^b	Yield ^c , %	Product	ratio ^d cis : trans
7a	PhMgBr	100	11a+11b	82 : 18
7a	CH ₃ MgBr	94	12a+12b	97 : 3
7b	PhMgBr	96	13	100 : 0
7b	CH ₃ MgBr	82	14	100 : 0

^aReactions were generally performed in THF at rt for 40 h. ^bThree equiv. of Grignard reagents were used. ^cYield of total product obtained after column chromatography. ^dRatio determined by ¹H-NMR.

The stereochemical course of the Grignard addition to the 1,3-oxazolidine, which is widely known as a masked imine derivative, has been well clarified. Similarly, the stereoselectivity of the Grignard addition to a piperidinium ion intermediate leading to α , α '-disubstituted piperidine presumably originates from a stereoselective effect. By way of explanation, we can examine the case of the Grignard addition to bicyclic **7a**, as shown in Figure 1. Due to strong $A^{(1,2)}$ strain¹⁹ between the α -phenyl group and the *N*-phenylethyl group of iminium ion **I-A**, the α -phenyl group occupies the axial position through half-chair conformation **I-B**. The Grignard reagent stereoelectronically prefers the axial attack ^{5,20,21} on **I-B** to give the *cis*-2,6-diphenylpiperidine

derivative 11a as the major product. The emergence of *trans* isomer 11b (minor product) is due to steric hindrance between the bulky α -phenyl group of I-B and the attacking Grignard reagent, which raises the possibility of Grignard attack from the opposite direction (equatorial attack) to a limited extent. This is consistent with the results of the Grignard addition to bicyclic 7b, the small α -methyl moiety of which provides less steric hindrance for the stereoelectronically preferred axial attack, and thus gives the *cis* isomer as the sole product (13 and 14).

Figure 1

2.2. Asymmetric reaction of synthon 10a, 10b with Grignard reagents

Based on the finding that the 1-aza-4-oxabicyclo[4.3.0]nonane synthon bearing an (S)-phenyl or (R)-methyl moiety at C-9 mainly gave the cis- α , α '-dialkylpiperidine derivatives in the nucleophilic Grignard addition, we applied a similar reaction to bicyclic compounds bearing C-9 moieties with opposite configurations; i.e., (R)-phenyl and (S)-methyl (10a and 10b, respectively), as depicted in Table 2. The diastereofacial addition of phenylmagnesium bromide to bicyclic 10a gave the *trans* isomer of the α , α '-diphenylpiperidine derivative 15 in good yield. Analogously, the addition of methylmagnesium bromide to bicyclic 10b gave a mixture of the cis isomer 17a and the *trans* isomer 17b in a ratio of 62:38, respectively, in very good yield. This previous result prompted us to confirm the relation between the nature of the functionality at C-9 and cis-trans isomerism by reacting 10a with methylmagnesium bromide to almost exclusively give the *trans* isomer 16 in moderate yield, and the reaction of 10b with phenylmagnesium bromide gave the *trans* isomer 16 as the major product in excellent yield.

The cis-trans isomeric outcome of these nucleophilic additions varies depending upon the nature of the functionality at the stereogenic center C-9. This phenomenon can be explained as shown in Figure 2. As in the mechanism described in Figure 1, the piperidinium intermediate II for Grignard addition to 10a takes a half-chair conformation with an α -phenyl group at the axial position. The Grignard reagent can then attack the piperidinium intermediate from either direction. The bulkiness of the α -phenyl substituent²² hinders the axial attack of the Grignard reagent (II-A), and prefers the less sterically hindered equatorial attack (II-B) to give

only the *trans* isomer 15. The steric hindrance that leads to the equatorial attack is also produced by the phenyl function of the Grignard reagent, as shown in the reaction of 10b with phenylmagnesium bromide to predominantly give the *trans* isomer 16. On the other hand, such hindrance is minimized for the stereoelectronically preferred axial attack of 10b by methylmagnesium bromide to give the *cis* isomer 17a as the major product.

Table 2. Stereoselective Reactions of 10a/10b with Grignard Reagents^a

Substrate	Grignard reagent	Yield, %	Product	ratio cis : trans
10a	Ph M gBr	91	15	0 : 100
10a	CH ₃ MgBr	86	13+16	7 : 93
10b	PhMgBr	96	12a+16	13 : 87
10b	CH ₃ MgBr	97	17a+17b	62 : 38
10b	CH ₃ CH ₂ MgBr	93	18a+18b	53 : 47
10b	H ₂ C=CHMgBr	94	19a+19b	24 : 76
10b	HC≡CMgBr	87	20a+20b	3 : 97

 $^{^{}m a}$ All conditions were generally similar to those mentioned in Table 1.

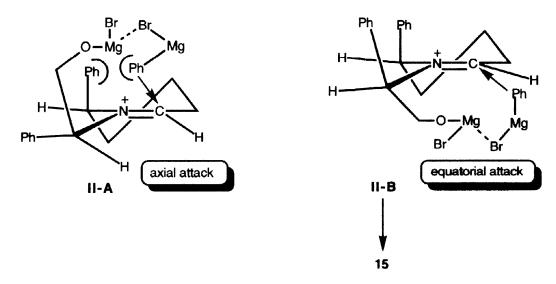


Figure 2

While the stereochemical outcome of the Grignard addition to 10a was unequivocally clarified as described above, 10b gave us a great opportunity to explore its reaction with various Grignard reagents. For the next experiment, we considered that despite the effect of the bulkiness of phenylmagnesium bromide, the C=C bond of the phenyl function of the Grignard reagent plays a predominant role in *cis-trans* isomerism. First, ethylmagnesium bromide was added to 10b to produce equal proportions of the *cis* isomer 18a and the *trans* isomer 18b in good yield. We then used the same procedure with vinylmagnesium bromide to give a 24:76 ratio of the *cis* isomer 19a and the *trans* isomer 19b in good yield. Under the same conditions, ethynylmagnesium bromide was used to give the *trans* isomer 20b almost exclusively and in moderate yield.

Chart 2

Surprisingly, however, there is a linear correlation between the number of bonds within the 2-membered alkyl function of the Grignard reagent and the *trans* isomer. Accordingly, the choice of Grignard reagent was found to be crucial for obtaining the *trans* isomer in good yield in these reactions. We do not yet know why Grignard reagents with double and triple bonds enhance the proportion of the *trans* isomer. The alkenyl and alkynyl character of Grignard reagents based on the HSAB principle, as suggested by Yamaguchi and coworkers, has been taken into consideration.

2.3. Determination of the stereochemistry of the resulting piperidine derivatives

We successfully prepared *cis*- and *trans*-2,6-dialkylpiperidine in a simple and concise manner from a single enantiomeric source. The piperidine derivatives obtained in this work were determined as follows. The absolute stereochemistries of 12a, 13 and 16, all of which are 2-methyl-6-phenylpiperidine derivatives that could be derived from two different bicyclic sources, were determined by cross-checking their NMR spectra. In addition, 12a and 13 were subjected to hydrogenolysis using a palladium hydroxide catalyst to give the enantiomeric pair 21, $[\alpha]^{24}_{D}$ -35.6 (c 1.38, CHCl₃), and 22, $[\alpha]^{24}_{D}$ +35.2 (c 1.26, CHCl₃), in moderate to good yield. Compound 16 was subjected to oxidative cleavage using lead tetraacetate in acetic acid to give the enantiomer 23, $[\alpha]^{24}_{D}$ +33.4 (c 1.25, CHCl₃), in moderate yield (Chart 2).

The stereochemistries of the *cis-trans* isomers 19a and 19b, which have a vinyl side chain, were established by NOE experiments, the more important details of which are summarized in Figure 3. With the stereochemistries of 19a and 19b in hand, we carried out the hydrogenation of 19b using a platinum oxide catalyst to provide a single enantiomer whose NMR spectra are consistent with those of the *trans* isomers 18b. Next, the hydrogenation of the *trans* isomers 20b using the Lindlar catalyst gave 19b.

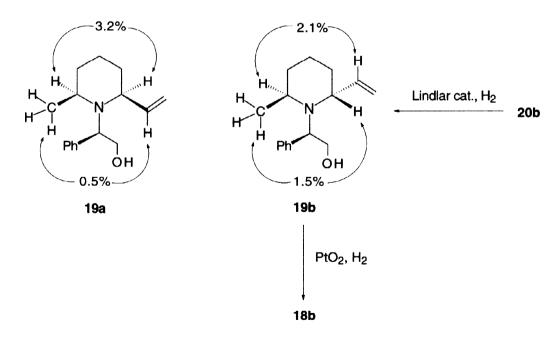


Figure 3

2.4. Enantioselective synthesis of (+)-solenopsin A (2b) and both enantiomers of isosolenopsin A (3a and 3b)

We applied our findinds to the synthesis of (-)-isosolenopsin A (3a), beginning with Grignard addition to bicyclic 7b to give the diasteromerically pure piperidine 24 in 98% yield (Scheme 2). After reductive cleavage of the chiral auxiliary over 5% palladium on carbon under a hydrogen atmosphere, followed by ethanolic-hydrochloric acidification, (-)-isosolenopsin A hydrochloride (3a·HCl) was obtained as colorless crystals, mp 152-153 °C (CH₂Cl₂-ether) (lit. 1 mp 154-155 °C); $[\alpha]^{24}_{D}$ +10.0 (c 1.1, CHCl₃) (lit. 3 (+)-isosolenopsin A hydrochloride: $[\alpha]^{20}_{D}$ -10.3 (c 1.3, CHCl₃)) in 82% yield.

Scheme 2

The same route was successfully used to prepare (+)-isosolenopsin A (3b) through the reaction of bicyclic 10b with the correct Grignard reagent to give a diastereomeric mixture, from which the desired *cis* isomer 27 was easily separated in 62% yield. A similar elimination of the *N*-functionality, and acidification of the resulting secondary amine gave (+)-isosolenopsin A hydrochloride (3b·HCl) as colorless crystals, mp 152–153 °C (CH₂Cl₂-ether) (lit. 1 mp 154–155 °C); $[\alpha]_{-}^{24}$ –10.1 (*c* 1.0, CHCl₃) (lit. 3 $[\alpha]_{-}^{20}$ –10.3 (*c* 1.3, CHCl₃)) in 86% yield.

Scheme 3

The synthesis of (+)-solenopsin A (2b) was achieved starting from 20b, which was derived as above from 10b. The 11-membered aliphatic side chain was built up by introducing readily available (*E*)-1-iodo-1-nonene²³ to 20b, using tetrakis(triphenylphosphine)palladium catalysis, ²⁴ to give the unsaturated piperidine 25 in 88% yield. The subsequent stage is critical because migration of the double bond during the palladium on carbon catalytic hydrogenation causes a loss of configuration. Based on the Wasserman procedure, ⁵ the unsaturated side chain was first hydrogenated with platinum oxide to give the saturated piperidine 26 in 93% yield, and this was followed by hydrogenolytic cleavage of the chiral appendage and salt formation to yield the (+)-solenopsin A hydrochloride (2b·HCl) as colorless crystals, mp 151–152 °C (CH₂Cl₂-ether) (lit. ¹⁰ mp 146 °C); $[\alpha]_{p}^{24} + 8.0$ (c 1.3, CHCl₃) (lit. ¹⁰ $[\alpha]_{p}^{20} + 7.5$ (c 1.3, CHCl₃)) in 87% yield.

3. CONCLUSION

We have developed a simple procedure for the enantioselective preparation of *cis*- and *trans*-2,6-dialkylpiperidine derivatives *via* Grignard addition to 1-aza-4-oxabicyclo[4.3.0]nonane derivatives. Although the mechanism that exclusively provides the *trans* isomer is still unclear, this procedure was successfully applied to the synthesis of (+)-solenopsin A (2b) in 29% overall yield in seven steps from the chiral auxiliary 4. We also achieved the enantioselective synthesis of (-)-isosolenopsin A (3a) in 49% overall yield in five steps, and of (+)-isosolenopsin A (3b) in 25% overall yield in five steps, from a single enantiomeric source (4). We are confident that this methodology may be applicable to the enantioselective synthesis of numerous 2,6-disubstituted piperidine alkaloids.

4. EXPERIMENTAL SECTION

General procedures

Melting points were measured without correction. The ¹H NMR and ¹³C NMR spectra were run in CDCl₃ unless otherwise noted. All chemical shifts are reported as δ values (ppm) relative to TMS and residual CDCl₃ as internal standards on 270 Mhz and 500 Mhz spectrometer. IR spectra were recorded on a JASCO FT/IR-200, and major absorptions are listed in cm⁻¹. Mass spectra and High-resolution mass spectra were recorded on a JEOL JMS 600 spectrometer in the chemical ionization (CI) with isobutane and electron impact (EI) methods. Optical rotations were performed on a JASCO DIP-1000; concentrations reported are in g/100 mL. Column chromatography was performed on silica gel (45~75 mm, Wakogel C-300). The THF was distilled over potassium metal. All other solvents and reactants were of the best commercial grade available and used without further purification unless noted.

Reaction of the bicyclic compounds with Grignard reagents

To a stirred solution of bicyclic compound (7a, 7b, 10a, 10b)¹⁴ in THF (10 mL) was added dropwise a solution of Grignard reagent (3 equiv). After being stirred at 10–20°C for 2 d, the reaction mixture was quenched with water and the organic solution was decanted from the insoluble solid. The residue was extracted with ether (2 x 10 mL), then the organic extracts were combined, dried over anhydrous Na₂SO₄ and concentrated on a rotary evaporator to give a crude product or a diastereomeric mixture of 2,6-disubstituted piperidine derivative (12a, 13, 16, 18a, 18b, 19a, 19b, 20b, 24, 27). The crude product or diastereomeric

mixture was subjected to column chromatography on silica gel using appropriate solvent as eluent to give the requisite compound in purity.

(2S,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-phenylpiperidine (12a)

A 3 mol/L solution of methylmagnesium bromide in THF was added to **7a** to give a diastereomeric mixture (97:3) of **12a+12b** in 94% yield. Separation by column chromatography on silica gel with CH_2Cl_2 –MeOH (50:1) gave **12a** and **12b**. Major product: **12a**, colorless crystals (from hexane), mp 77–78 °C. $[\alpha]^{24}_{D}$ –162.8 (*c* 1.13, CHCl₃). ¹H NMR δ 1.26–1.48 (m, 3 H), 1.27 (d, 3 H, J = 6.4 Hz), 1.60–1.76 (m, 3 H), 2.83 (m, 1 H), 3.46 (br s, 1 H), 3.52 (dd, 1 H, J = 2.5, 6.3 Hz), 3.86 (t, 1 H, J = 7.2 Hz), 3.93–4.04 (m, 2 H), 7.24–7.47 (m, 10 H). ¹³C NMR δ 21.18, 23.83, 32.40, 35.20, 50.48, 61.76, 62.03, 64.82, 127.17, 127.39, 127.82, 128.13, 128.64, 128.77, 136.99, 145.16. EIMS m/z (relative intensity): 295 [M]⁺ (3), 264 [M – CH_2OH]⁺ (100). IR (CHCl₃): 3440 (OH) cm⁻¹. Anal. Calcd for $C_{20}H_{25}NO$: C, 81.31; H, 8.53; N, 4.74. Found: C, 81.43; H, 8.66; N, 4.78.

(2R,6R,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-phenylpiperidine (13)

A 1 mol/L solution of phenylmagnesium bromide in THF was added to **7b** to give a single diastereomer of **13** in 96% yield. Purification by column chromatography on silica gel with CH_2Cl_2 –MeOH (50:1) gave the pure **13** as an almost colorless oil. $[\alpha]_D^{24}$ +49.7 (c 1.15, CHCl₃). H NMR δ 1.22–1.78 (m, 6 H), 1.23 (d, 3 H, J = 5.9 Hz), 2.69 (br s, 1 H), 3.07 (m, 1 H), 3.38 (t, 1 H, J = 9.4 Hz), 3.50–3.58 (m, 2 H), 4.28 (dd, 1 H, J = 6.3, 9.4 Hz), 7.12–7.38 (m, 10 H). CNMR δ 23.11, 23.72, 35.38, 36.86, 55.06, 60.64, 62.14, 62.77, 126.93, 127.01, 127.63, 127.74, 127.96, 128.57, 139.30, 145.69. EIMS m/z (relative intensity): 295 [M]⁺ (3), 264 [M – CH_2OH]⁺ (100). IR ($CHCl_3$): 3400 (OH) cm⁻¹. Anal. Calcd for $C_{20}H_{25}NO$: C, 81.31; H, 8.53; N, 4.74. Found: C, 81.15; H, 8.52; N, 4.75.

(2S,6R,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-phenylpiperidine (16)

A 3 mol/L solution of methylmagnesium bromide in THF was added to **10a** to give a diastereomeric mixture (7:93) of **13+16** in 86% yield. Separation by column chromatography on silica gel with CH_2Cl_2 –MeOH (50:1) gave **13** and **16**. Major product: **16**, colorless oil. [α]²⁴_D –92.2 (c 1.31, CHCl₃). ¹H NMR δ 1.18–1.31 (m, 2 H), 1.29 (d, 3 H, J = 6.9 Hz), 1.49–1.75 (m, 4 H), 2.38 (br s, 1 H), 3.34 (m, 1 H), 3.54 (m, 1 H), 3.83–3.93 (m, 2 H), 4.29 (br d, 1 H, J = 8.9 Hz), 7.23–7.44 (m, 10 H). ¹³C NMR δ 18.48, 20.11, 29.96, 30.28, 47.79, 57.38, 60.68, 60.78, 126.94, 127.21, 128.19, 128.24, 128.48, 128.68, 140.77, 143.17. EIMS m/z (relative intensity): 295 [M]⁺ (3), 264 [M – CH₂OH]⁺ (100). IR (CHCl₃): 3390 (OH) cm⁻¹. Anal. Calcd for $C_{20}H_{25}NO$: C, 81.31; H, 8.53; N, 4.74. Found: C, 81.17; H, 8.53; N, 4.58.

(2R,6S,1'R)- and (2S,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-ethyl-6-methylpiperidine (18a and 18b)

A 1 mol/L solution of ethylmagnesium bromide in THF was added to **10b** to give a diastereomeric mixture (53:47) of **18a+18b** in 93% yield. Separation by column chromatography on silica gel with CH_2Cl_2 -MeOH (50:1) gave **18a** and **18b**. Major product: **18a**, colorless oil. $[\alpha]_D^{24}$ -18.8 (c 1.09, $CHCl_3$). H NMR δ 0.73 (t, 3 H, J = 7.4 Hz), 1.13 (d, 3 H, J = 6.9 Hz), 1.22–1.72 (m, 8 H), 2.68 (m, 1 H), 2.94 (br s, 1 H), 3.16 (m, 1 H), 3.72 (dd, 1 H, J = 5.6, 10.4 Hz), 3.86 (dd, 1 H, J = 7.6, 10.4 Hz), 4.01 (dd, 1 H, J = 5.6, 7.6 Hz), 7.24–7.34 (m, 5 H). C NMR δ 11.90, 15.80, 20.66, 26.10, 27.54, 30.43, 47.65, 58.27, 61.83, 65.93,

127.47, 128.29, 128.38, 140.20. EIMS m/z (relative intensity): 247 [M]⁺ (3), 216 [M - CH₂OH]⁺ (44). IR (CHCl₃): 3420 (OH) cm⁻¹. Anal. Calcd for C₁₆H₂₅NO: C, 77.68; H, 10.19; N, 5.66. Found: C, 77.71; H, 10.30; N, 5.68. Minor product: **18b**, colorless oil. $[\alpha]^{24}_{D}$ -19.9 (c 1.27, CHCl₃). ¹H NMR δ 0.77-1.12 (m, 4 H), 0.95 (t, 3 H, J = 7.4 Hz), 1.27 (d, 3 H, J = 6.9 Hz), 1.31-1.77 (m, 4 H), 2.99 (m, 1 H), 3.33 (m, 1 H), 3.46 (dd, 1 H, J = 5.6, 10.4 Hz), 3.86 (t, 1 H, J = 10.4 Hz), 4.20 (dd, 1 H, J = 5.6, 10.4 Hz), 7.24-7.39 (m, 5 H). ¹³C NMR δ 11.83, 20.20, 20.60, 25.48, 25.93, 29.90, 48.52, 54.10, 58.96, 60.33, 127.47, 128.29, 129.17, 141.20. EIMS m/z (relative intensity): 247 [M]⁺ (9), 216 [M - CH₂OH]⁺ (100). IR (CHCl₃): 3430 (OH) cm⁻¹. Anal. Calcd for C₁₆H₂₅NO: C, 77.68; H, 10.19; N, 5.66. Found: C, 77.66; H, 10.26; N, 5.50.

(2S,6S,1'R)- and (2R,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-ethenyl-6-methylpiperidine (19a and 19b)

A 1 mol/L solution of vinylmagnesium bromide in THF was added to **10b** to give a diastereomeric mixture (24:76) of **19a+19b** in 94% yield. Separation by column chromatography on silica gel with CH_2CI_2 –EtOAc (5:1) gave **19a** and **19b**. Major product: **19b**, colorless oil. [α]²⁴_D –20.2 (c 1.30, CHCl₃). ¹H NMR δ 1.08–1.34 (m, 4 H), 1.21 (d, 3 H, J = 6.8 Hz), 1.42–1.51 (m, 2 H), 2.85 (br s, 1 H), 3.30 (m, 1 H), 3.66 (dd, 1 H, J = 6.1, 10.6 Hz), 3.70 (m, 1 H), 3.89 (dd, 1 H, J = 9.1, 10.6 Hz), 4.24 (dd, 1 H, J = 6.1, 9.1 Hz), 5.13–5.20 (m, 2 H), 6.07 (ddd, 1 H, J = 5.9, 9.4, 11.2 Hz), 7.23–7.38 (m, 5 H). ¹³C NMR δ 19.62, 19.74, 28.57, 30.84, 48.46, 55.45, 60.42, 61.34, 115.21, 127.32, 128.25, 128.91, 140.79, 141.01. EIMS m/z (relative intensity): 245 [M]⁺ (4), 214 [M – CH₂OH]⁺ (100). IR (CHCl₃): 3400 (OH) cm⁻¹. Anal. Calcd for $C_{16}H_{23}$ NO: C, 78.32; H, 9.45; N, 5.71. Found: C, 78.07; H, 9.53; N, 5.49. Minor product: **19a**, colorless oil. [α]²⁴_D –71.9 (c 1.05, CHCl₃). ¹H NMR δ 1.06 (d, 3 H, J = 6.3 Hz), 1.20–1.69 (m, 6 H), 2.61 (m, 1 H), 3.20 (br s, 1 H), 3.44 (dt, 1 H, J = 2.6, 8.9 Hz), 3.75 (dd, 1 H, J = 6.3, 10.1 Hz), 3.92 (t, 1 H, J = 10.1 Hz), 4.44 (dd, 1 H, J = 6.3, 10.1 Hz), 5.10–5.22 (m, 2 H), 5.87 (ddd, 1 H, J = 8.9, 10.1, 17.3 Hz), 7.24–7.38 (m, 5 H). ¹³C NMR δ 21.45, 21.94, 29.65, 33.77, 34.57, 51.33, 61.89, 64.75, 115.67, 127.17, 128.09, 128.14, 138.49, 142.65. EIMS m/z (relative intensity): 245 [M]⁺ (3), 214 [M – CH₂OH]⁺ (100). IR (CHCl₃): 3420 (OH) cm⁻¹. Anal. Calcd for $C_{16}H_{23}$ NO: C, 78.32; H, 9.45; N, 5.71. Found: C, 77.98; H, 9.63; N, 5.52.

(2R,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-ethynyl-6-methylpiperidine (20b)

A 0.5 mol/L solution of ethynylmagnesium bromide in THF was added to **10b** to give a diastereomeric mixture (3:97) of **20a+20b** in 87% yield. Separation by column chromatography on silica gel with hexane–EtOAc (5:1) gave **20a** and **20b**. Major product: **20b**, pale yellow oil. $[\alpha]^{24}_{D}$ +122.5 (c 1.22, CHCl₃). ¹H NMR δ 1.21 (d, 3 H, J = 6.3 Hz), 1.48–1.85 (m, 6 H), 2.39 (d, 1 H, J = 2.0 Hz), 3.35 (m, 1 H), 3.70 (m, 1 H), 4.26–4.67 (m, 3 H), 7.22–7.45 (m, 5 H). ¹³C NMR δ 20.72, 21.30, 31.77, 36.09, 45.57, 50.48, 61.06, 61.38, 73.24, 85.61, 126.86, 128.13, 128.22, 140.95. EIMS m/z (relative intensity): 243 [M]⁺ (4), 212 [M – CH₂OH]⁺ (100). IR (CHCl₃): 3400 (OH), 3300 (C≡CH) cm⁻¹. Anal. Calcd for C₁₆H₂₁NO: C, 78.97; H, 8.70; N, 5.76. Found: C, 78.79; H, 8.56; N, 5.69.

(2R,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-undecylpiperidine (24)

A solution of undecylmagnesium bromide in THF, prepared *in situ* from Mg turnings and 1-bromoundecane, was added to **7b** to give a single diastereomer of **24**. Purification by column chromatography on silica gel with CH_2Cl_2 -EtOAc (1:1) gave **24** as a colorless oil in 98% yield. [α]²⁴_D+10.2 (c 1.1, CHCl₃). ¹H NMR δ 0.88 (t, 3 H, J= 7.3 Hz), 1.06 (d, 3 H, J= 6.1 Hz), 1.13–1.62 (m, 10 H), 1.26 (br s, 16 H), 2.83 (m,

1 H), 2.95 (br s, 1 H), 3.01 (m, 1 H), 3.70 (dd, 1 H, J = 5.5, 10.4 Hz), 3.79 (dd, 1 H, J = 7.3, 10.4 Hz), 4.00 (dd, 1 H, J = 5.5, 7.3 Hz), 7.26–7.33 (m, 5 H). ¹³C NMR δ 14.09, 15.66, 20.10, 22.66, 26.20, 28.13, 29.33, 29.62 (4 C), 29.86, 31.14, 31.89, 34.27, 50.59, 53.07, 61.76, 65.14, 127.47, 128.28, 128.43, 139.90. EIMS m/z (relative intensity): 373 [M]⁺ (7), 342 [M – CH₂OH]⁺ (79). IR (CHCl₃): 3400 (OH) cm⁻¹. Anal. Calcd for $C_{25}H_{43}NO$: C, 80.37; H, 11.60; N, 3.75. Found: C, 80.19; H, 11.75; N, 3.62.

(2S,6R,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-undecylpiperidine (27)

A solution of readily available undecylmagnesium bromide in THF was added to **10b** to give a pale yellow oil as a diastereomeric mixture of **26+27**. Separation and purification by column chromatography on silica gel with CH₂Cl₂-EtOAc (1:1) gave **27** as a colorless oil in 62% yield. [α]²⁴_D -41.2 (c 2.28, CHCl₃). ¹H NMR δ 0.88 (t, 3 H, J = 6.7 Hz), 0.96 (m, 1 H), 1.13 (d, 3 H, J = 6.7 Hz), 1.21–1.62 (m, 15 H), 1.26 (br s, 10 H), 2.74 (m, 1 H), 3.16 (m, 1 H), 3.72 (dd, 1 H, J = 5.5, 10.4 Hz), 3.83 (dd, 1 H, J = 7.3, 10.4 Hz), 3.99 (dd, 1 H, J = 5.5, 7.3 Hz), 7.25–7.33 (m, 5 H). ¹³C NMR δ 14.11, 15.71, 20.47, 22.67, 27.78, 28.14, 29.34, 29.59 (2 C), 29.63 (2 C), 29.82, 30.46, 31.90, 33.42, 47.82, 56.82, 61.89, 66.02, 127.53, 128.32, 128.46, 140.06. EIMS m/z (relative intensity): 373 [M]⁺ (5), 342 [M – CH₂OH]⁺ (22). IR (CHCl₃): 3400 (OH) cm⁻¹. HRMS Calcd for C₂₅H₄₃NO: 373.3347. Found: 373.3354.

(2S,6S)-2-Methyl-6-phenylpiperidine (21)

A solution of **12a** in MeOH (10 mL) was hydrogenated under 1 atm pressure of hydrogen with palladium hydroxide on carbon (Pearlman's catalyst) at rt for 12 h. After the catalyst was removed through Celite, the organic solution was concentrated in vacuo to give a pale yellow oil. Filtration through silica gel with CH_2Cl_2 –MeOH (20:1) gave piperidine **21** as a colorless oil in 97% yield. [α]²⁰_D –35.6 (c 1.38, CHCl₃). ¹H NMR δ 1.11 (d, 3 H, J = 6.3 Hz), 1.17 (m, 1 H), 1.38–1.91 (m, 6 H), 2.81 (m, 1 H), 3.66 (dd, 1 H, J = 2.6, 10.6 Hz), 7.20–7.40 (m, 5 H). ¹³C NMR δ 23.01, 25.33, 33.78, 34.18, 53.15, 62.44, 126.68, 126.92, 128.28, 145.42. EIMS m/z (relative intensity): 175 [M]⁺ (43), 160 [M – CH₃]⁺ (100). HRMS Calcd for $C_{25}H_{43}$ NO: 175.1361. Found: 175.1344.

(2R,6R)-2-Methyl-6-phenylpiperidine (22)

A solution of **13** in MeOH (10 mL) was hydrogenated under 1 atm pressure of hydrogen with palladium hydroxide on carbon (Pearlman's catalyst) at rt for 12 h. After the catalyst was removed through Celite, the organic solution was concentrated in vacuo to give a pale yellow oil. Filtration through silica gel with CH_2Cl_2 –MeOH (20:1) gave the secondary amine **22** as a colorless oil in 85% yield. [α]²⁰_D +35.2 (c 1.26, CHCl₃). ¹H NMR δ 1.11 (d, 3 H, J = 6.3 Hz), 1.17 (m, 1 H), 1.38–1.91 (m, 6 H), 2.81 (m, 1 H), 3.65 (dd, 1 H, J = 2.5, 10.5 Hz), 7.20–7.39 (m, 5 H). ¹³C NMR δ 23.01, 25.32, 33.76, 34.16, 53.15, 62.44, 126.68, 126.93, 128.29, 145.40. EIMS m/z (relative intensity): 175 [M]⁺ (42), 160 [M – CH₃]⁺ (100). HRMS Calcd for $C_{25}H_{43}$ NO: 175.1361. Found: 175.1371.

(2S,6R)-2-Methyl-6-phenylpiperidine (23)

To a solution of 16 in glacial acetic acid (10 mL) was added lead tetraacetate (1.5 equivalent). After the reaction mixture was stirred at 60°C for 14 h, it was quenched with the addition of water (20 mL) and 1 N sodium hydroxide solution (40 mL), respectively. The resulted basic mixture was extracted with ether (3 x 15

mL), the organic extracts were combined and concentrated *in vacuo* to give a yellow oil. The crude oil was subjected to column chromatography on silica gel with CH_2Cl_2 –MeOH (20:1) to give **23** as a colorless oil in 68% yield. [α]²⁰_D +33.4 (c 1.25, CHCl₃). ¹H NMR δ 1.21 (d, 3 H, J = 6.6 Hz), 1.40 (m, 1 H), 1.56–1.87 (m, 6 H), 3.29 (m, 1 H), 4.04 (dd, 1 H, J = 3.7, 7.5 Hz), 7.18–7.39 (m, 5 H). ¹³C NMR δ 19.77, 19.90, 31.28, 33.27, 47.24, 54.10, 126.60, 126.68, 128.28, 145.05. EIMS m/z (relative intensity): 175 [M]⁺ (36), 160 [M – CH₃]⁺ (100). HRMS Calcd for $C_{25}H_{43}$ NO: 175.1361. Found: 175.1360.

(2S,6R,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-[(E)-undec-3-en-1-ynyl]piperidine (25)

To a stirred solution of (*E*)-1-iodo-1-nonene (1.77 g, 7.0 mmol) and tetrakis(triphenylphosphine)-palladium (400 mg, 0.346 mmol) in pyrrolidine (5 mL), under an argon atmosphere, was added a solution of **20b** (1.42 g, 5.84 mmol) in pyrrolidine (5 mL). After stirring at rt for 12 h, the mixture was hydrolyzed with a saturated aqueous solution of ammonium chloride and extracted with ether. The organic extract was dried over anhydrous Na₂SO₄ and concentrated on a rotary evaporator. Filtration through silica gel with hexane–EtOAc (10:1) gave **25** as a pale yellow oil (1.88 g, 88%). $\left[\alpha\right]_{D}^{24}$ +213.8 (*c* 1.28, CHCl₃). H NMR δ 0.89 (t, 3 H, J = 6.9 Hz), 1.20 (d, 3 H, J = 6.4 Hz), 1.19–1.62 (m, 8 H), 1.28 (br s, 8 H), 1.76 (m, 1 H), 2.03 (br s, 1 H), 2.10 (m, 1 H), 3.35 (m, 1 H), 3.76 (m, 1 H), 4.25–4.36 (m, 3 H), 5.51 (ddd, 1 H, J = 1.6, 3.3, 15.8 Hz), 6.14 (dt, 1 H, J = 7.0, 15.8 Hz), 7.21–7.36 (m, 3 H), 7.44 (d, 2 H, J = 7.4 Hz). The NMR δ 14.04, 20.95, 21.26, 22.60, 28.66, 29.07, 29.08, 31.72, 32.06, 33.02, 36.19, 46.26, 50.45, 61.17, 61.50, 84.33, 89.16, 108.99, 126.75, 128.13, 128.17, 141.02, 144.47. EIMS m/z (relative intensity): 367 [M]* (1), 336 [M – CH₂OH]* (100). IR (CHCl₃): 3410 (OH) cm⁻¹. Anal. Calcd for C₂₅H₃₇NO: C, 81.69; H, 10.15; N, 3.81. Found: C, 81.52; H, 10.18; N, 3.75.

(2S,6S,1'R)-N-(2-Hydroxy-1-phenylethyl)-2-methyl-6-undecylpiperidine (26)

A solution of **25** (0.40 g, 1.088 mmol) in benzene (10 mL) and glacial acetic acid (0.02 mL) was hydrogenated under 1 atm pressure of hydrogen with platinum oxide catalyst (15 mg) at rt for 3 h. After the catalyst was removed through Celite, the organic solution was concentrated on a rotary evaporator to give a pale yellow oil. Purification by column chromatography on silica gel with hexane–EtOAc (10:1) gave **26** as a colorless oil (0.38 g, 93%). [α]²⁴_D –1.5 (c 1.52, CHCl₃). ¹H NMR δ 0.72–1.12 (m, 4 H), 0.88 (t, 3 H, J = 6.7 Hz), 1.14–1.74 (m, 12 H), 1.23 (d, 3 H, J = 6.7 Hz), 1.28 (s, 10 H), 3.06 (m, 1 H), 3.32 (m, 1 H), 3.43 (dd, 1 H, J = 5.5, 10.4 Hz), 3.85 (t, 1 H, J = 10.4 Hz), 4.18 (dd, 1 H, J = 5.5, 10.4 Hz), 7.26–7.34 (m, 5 H). ¹³C NMR δ 14.10, 20.27, 20.66, 22.67, 26.37, 27.43, 29.34, 29.66 (4 C), 29.89 (2 C), 31.90, 33.02, 48.50, 52.50, 58.96, 60.32, 127.48, 128.31, 129.21, 141.29. EIMS m/z (relative intensity): 373 [M]⁺ (8), 342 [M – CH₂OH]⁺ (45). IR (CHCl₃): 3400 (OH) cm⁻¹. Anal. Calcd for C₂₅H₄₃NO: C, 80.37; H, 11.60; N, 3.75. Found: C, 80.28; H, 11.60; N, 3.62.

(2R,6S)-(-)-Isosolenopsin A hydrochloride (3a·HCl)

To a solution of 24 in MeOH (10 mL) was added 5% palladium on carbon, and hydrogenated under 1 atm pressure of hydrogen at rt for 12 h. The reaction mixture was filtered through Celite, and the filtrate was concentrated after addition of a few drops of ethanolic–HCl. To the resultant white solid was added ether, the insoluble part was separated and recrystallized from CH_2Cl_2 -ether to give (-)-isosolenopsin A hydrochloride (3a·HCl) as colorless crystals in 82% yield, mp 152–153 °C. [α]²⁴_D+10.0 (c 1.1, CHCl₃). ¹H NMR δ 0.88 (t,

3 H, J = 6.7 Hz), 1.24 (br s, 18 H), 1.58 (d, 3 H, J = 6.7 Hz), 1.64 (br s, 3 H), 1.74–1.99 (m, 4 H), 2.16 (m, 1 H), 2.89 (m, 1 H), 3.07 (m, 1 H), 9.06 (br s, 1 H), 9.44 (br s, 1 H). ¹³C NMR δ 14.12, 22.68, 23.11, 24.89, 26.00, 29.35, 29.60 (3 C), 29.66, 29.84, 31.91, 32.28, 34.44, 37.46, 52.48, 57.14. EIMS m/z (relative intensity): 253 [M]⁺ (18), 238 [M – CH₃]⁺ (61).

(2S,6R)-(+)-Isosolenopsin A hydrochloride (3b·HCl)

To a solution of **27** in MeOH (10 mL) was added 5% palladium on carbon, and hydrogenated under 1 atm pressure of hydrogen at rt for 12 h. After work-up and recrystallization procedure in the same manner as described for **3a**·HCl, the (+)-isosolenopsin A hydrochloride (**3b**·HCl) was obtained as colorless crystals in 86% yield, mp 152–153 °C. $[\alpha]^{24}_D$ –10.1 (c 1.0, CHCl₃). ¹H NMR δ 0.88 (t, 3 H, J = 6.7 Hz), 1.24 (br s, 18 H), 1.57 (d, 3 H, J = 6.7 Hz), 1.71 (br s, 3 H), 1.74–1.99 (m, 4 H), 2.14 (m, 1 H), 2.89 (m, 1 H), 3.08 (m, 1 H), 9.02 (br s, 1 H), 9.42 (br s, 1 H). ¹³C NMR δ 14.11, 22.68, 23.09, 24.87, 25.99, 29.34, 29.60 (3 C), 29.66, 29.83, 31.91, 32.26, 34.42, 37.44, 52.49, 57.14. EIMS m/z (relative intensity): 253 [M]⁺ (22), 238 [M – CH₃]⁺ (65).

(2S,6S)-(+)-Solenopsin A hydrochloride $(2b \cdot HCl)$

To a solution of **26** in MeOH (10 mL) was added 5% palladium on carbon, and hydrogenated under 1 atm pressure of hydrogen at rt for 12 h. After work-up and recrystallization procedure in the same manner as described for **3a**·HCl, the (+)-solenopsin A hydrochloride (**2b**·HCl) was obtained as colorless crystals in 87% yield, mp 151–152 °C. [α]²⁴_D +8.0 (c 1.3, CHCl₃). ¹H NMR δ 0.88 (t, 3 H, J = 6.7 Hz), 1.25 (br s, 19 H), 1.48 (d, 3 H, J = 6.7 Hz), 1.58–1.82 (m, 4 H), 1.87–2.18 (m, 3 H), 3.27 (m, 1 H), 3.54 (m, 1 H), 9.35 (br s, 2 H). ¹³C NMR δ 14.08, 19.53, 21.21, 22.66, 26.43, 29.32, 29.60 (3 C), 29.63, 29.76, 30.75, 31.88, 32.95, 34.03, 45.79, 50.80. EIMS m/z (relative intensity): 253 [M]⁺ (9), 238 [M – CH₃]⁺ (72).

5. ACKNOWLEDGMENT

The authors are grateful to Mrs. Toshiko Ogata for the elemental analyses, Mrs. Yoshiko Kawada for the NOE experiments, and Mr. Hideaki Komiya and Mrs. Hiroko Kasai for the mass spectral measurements. The Ministry of Education, Science, Sports and Culture as well as the Inpex Foundation of Japan are gratefully acknowledged for granting scholarships to H. P.

6. REFERENCES

- 1. MacConnell, J. G.; Blum, M. S.; Fales, H. M. Tetrahedron 1971, 27, 1129-1139.
- 2. Jones, T. H; Blum, M. S.; Fales, H. M. ibid. 1982, 38, 1949-1958.
- 3. Leclercq, S.; Thirionet, I.; Broeders, F.; Daloze, D.; Vander Meer, R.; Braekman, J. C. *ibid.* **1994**, *50*, 8465-8478.
- 4. Mundy, B. P.; Bjorklund, M. Tetrahedron Lett. 1985, 26, 3899-3902.
- 5. Wasserman, H. H.; Rusiecki, V. Tetrahedron Lett. 1988, 29, 4977-4980.
- 6. Ryckman, D. M.; Stevens, R. V. J. Org. Chem. 1987, 52, 4274-4279.
- 7. Jefford, C. W.; Wang, J. B. Tetrahedron Lett. 1993, 34, 2911-2914.

- 8. Yamaguchi, R.; Nakazono, Y.; Matsuki, T.; Hata, E.; Kawanisi, M. Bull. Chem. Soc. Jpn. 1987, 60, 215-222.
- 9. Meyers, A. I.; Edwards, P. D.; Bailey, T. R.; Jagdmann, G. E. J. Org. Chem. 1985, 50, 1019-1026.
- 10. Grierson, D. S.; Royer, J.; Guerrier, L.; Husson, H.-P. J. Org. Chem. 1986, 51, 4475-4477.
- 11. Comins, D. L.; Benjelloun, N. R. Tetrahedron Lett. 1994, 35, 829-832.
- 12. Maruoka, K.; Miyazaki, T.; Ando, M.; Matsumura, Y.; Sakane, S.; Hattori, K.; Yamamoto, H. J. Am. Chem. Soc. 1983, 105, 2831-2843.
- 13. Ogawa, M.; Natsume, M. Heterocycles 1984, 21, 769.
- 14. Poerwono, H.; Higashiyama, K.; Yamauchi, T.; Takahashi, H. Heterocycles 1997, 46, 385-400.
- 15. Poerwono, H.; Higashiyama, K.; Takahashi, H. ibid. 1998, 47, 263-270.
- 16. Poerwono, H.; Higashiyama, K.; Takahashi, H. J. Org. Chem. 1998, 63, 2711-2714.
- 17. Takahashi, H.; Hsieh, B.-C.; Higashiyama, K. Chem. Pharm. Bull. 1990, 38, 2429-2434.
- 18. Enders, D.; Reinhold, U. Tetrahedron: Asymmetry 1997, 8, 1895-1946.
- 19. Johnson, F. Chem. Rev. 1968, 68, 375-413.
- 20. Deslongchamps, P. Stereoelectronic Effects in Organic Chemistry; Pergamon Press: Oxford, 1983; pp. 209-290.
- 21. Comins, D. L.; Foley, M. A. Tetrahedron Lett. 1988, 29, 6711-6714.
- 22. Williams, R. M.; Sinclair, P. J.; Zhai, D.; Chen, D. J. Am. Chem. Soc. 1988, 110, 1547-1557.
- 23. Stille, J. K.; Simpson, J. H. J. Am. Chem. Soc. 1987, 109, 2138-2152.
- 24. Alami, M.; Ferri, F.; Linstrumelle, G. Tetrahedron Lett. 1993, 34, 6403-6406.